## MARK SCHEME for the October/November 2010 question paper

## for the guidance of teachers

## **5129 COMBINED SCIENCE**

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Pa	ge 2	Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL - October/November 2010	5129	02
1	(a)	0.18 (ign	ore units)		[1]
	(b)	V = <i>IR</i> o = 1.5	or 50 × 0.03 or 10 × 0.15 (V)		[2]
	(c)	Q = <i>It</i> <b>o</b> = 45 C	r <i>C</i> = <i>It</i> <b>or</b> 0.15 × 300		[3]
		0.75 C ga unit marl	ains 2 marks, 0.75 gains 1 mark < is independent of the numerical answer		
2	(a)	(i) blue	/ purple / indigo / violet		[1]
		<b>(ii)</b> OH⁻ igno	/ hydroxide ion re OH		[1]
	(b)	pipette burette ( neutral /	do not accept biuret) neutralised		[3]
	(c)	<b>(i)</b> (NH	4) <sub>2</sub> SO <sub>4</sub>		[1]
		(ii) fertil	iser		[1]
3	(a)	v = d/t or 3125 (m/ allow 2.7 allow 2.7	r speed = distance / time <b>or</b> 2.7 × 10 <sup>8</sup> / 24 × 60 × 60 (s) / × 10 <sup>8</sup> /24 = 11 250 000 for 1 mark / × 10 <sup>8</sup> / (24 × 60) = 187 500 for 1 mark		[2]
	(b)	F = ma <b>c</b> = 0.225 (	or a = F/m or 45/200 [m/s <sup>2</sup> ]		[2]
4	(a)	anther / s sepal ovary / c	stamen (ignore pollen grains) arpel		[3]
	(b)	to attract for pollin	i <u>nsects</u> ation		[2]
	(c)	anther /	stamen / X		[1]

	Page 3		Mark Scheme: Teachers' version Syllabus	Paper
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5	(a)   (	liquid – ir gas – rar	rregular shape majority of particles touching ndom particles not touching	[2]
	<b>(b)</b> r	melting condens	ation	[2]
6	(a) v I I I I I I I I I I I I I I I I I I I	volume density length resistanc colour e.m.f. pressure	ce / resistivity } any 2	[2]
	(b) s c r t t	smaller r constricti retains re triangula narrow b	range ion eading ar cross section oore / tube	
	r (	more ser (ignore n	nsitive J nore accurate / narrower alone)	[2]
	(C) r (	mercury <b>or</b> alcoho (ignore s	would freeze / would be solid ol stays liquid / does not freeze statement that mercury melts at –39°C)	[1]
7	(a) (	(i) tubir	ng	[1]
	(	ii) the v	water (in the beaker)	[1]
	(i	ii) the s do n	starch <b>1ot accept</b> starch and amylase	[1]
	(b) ( ;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;;	(amylase catalyses starch is sugar / n tube is <u>p</u> (allow co	e is an) enzyme s (breakdown of starch) broken down / digested naltose <u>diffuses</u> into the water <u>ermeable</u> (to maltose / sugar) prrect description for diffusion)	[3]
8	(a) /	E = Pt <b>o</b> = 216 J 3600 J ga	<b>r</b> energy = power × time <b>or</b> 1800 × 120 000 ains 2 marks, 3600 gains 1 mark	[3]
	<b>(b)</b> r	neutral earth AN	IY order	[2]

	Page 4				Mar	k Schen	ne: Teac	hers'	version		Syllabu	JS	Paper
				G		_EVEL -	- Octobe	er/Nove	ember 2010		5129		02
9	(a)	oxy	gen										[1]
	(b)	hyd	roger	ı									[1]
	(c)	hyd	roger	ı									[1]
	(d)	cart	oon m	nonoxie	de								[1]
	(e)	argo	on										[1]
10	(a)	(i)	N	S									[1]
		(ii)	S	N									[1]
	(b)	curr mag	ent n gnetic	ot cha c field r	nging / not chai	is consta nging / is	ant / in or s constar	ne dire nt	ction only				[2]
11	(a)	two (ger ferti allo	pare netica lisatio w cor	nts ally) dif on / fus overse	ferent c sion of g argume	offspring gametes ent	or nucle	i					[2]
	(D)												
		pen	is		••••••••••	•.		ca	rries sperm	and al	so urine		
		pros	state	gland	$\overline{\}$	·····		ca	rries sperm	but no	t urine		
		spe	rm dı	uct		$\checkmark$	/	····· all	ows sperm	to be r	eleased		
		test	is			$\square$	$\overline{}$	— pr	oduces spei	rm cell	S		
		uret	hra					∖ <sub>s∈</sub>	cretes semi	inal flui	d		

	Ра	ge 5		Mark Scheme: Tea	Syllabus	Paper	
			GCE	O LEVEL – Octob	er/November 2010	5129	02
12	(a)	54 5.4 (acce	ot both above	112 11.2 divided by 10) 5.6		[2] [1]	
		(acce (corr	ot above divid ct answers a	ded by 2) Iways gain credit)			[.]
	(b)	oxida	ion / redox				[1]
13	(a)	46°					[1]
	(b)	(i) I	.I. = sin <i>i</i> / sir	ı <i>r</i>			[1]
		(ii) 2	3° (accept 27	.79 to 28)			[1]
14	(a)	gas / gas l liquic gas l	= hydroge = carbon o C = water = water / s	n dioxide steam			[4]
	(b)	ethei	e contains a	(carbon to carbon) o	double bond		[1]
	(c)	spee	up the react	ion / lowers activation	on energy		[1]
15	(a)	(i)	ss of water				[1]
		<b>(ii)</b> t i	rough stoma leaves	ta			[2]
	(b)	(i) (i	hours 1 hours (acce	ept 23.5 to 24)			[2]
		<b>(ii)</b> i	will wilt / droo	qq			[1]
16	(a)	cond	ction / condu	ct / conductor			[1]
	(b)	it has	expanded				[1]

	Page 6			Mark Scheme: Teachers' version	Syllabus	Paper
				GCE O LEVEL – October/November 2010	5129	02
17	(a)	(i)	8 10			[2]
		(ii)	2 ele	ectrons on inner ring and 6 electrons on outer ring		[1]
	(b)	sar diff (igr	ne ele erent nore r	ement / same number of protons number of neutrons / nucleon number eferences to electrons)		[2]
	(c)	oxy (ox oxy ste ign ans	vgen t y-ace vgen t el ma ore bi swers	ents in hospital tylene) welding anks for divers nufacture reathing / saving lives / respiration / combustion which relate to breathing must say 'how' or 'what' is do	one	[2]
18	bloo glaı targ live	od nd get o r	rgan			[4]
19	(a)	d g/c (co (un	= m / = 3.0 m <sup>3</sup> rrect a it mar	v <b>or</b> 5.4 / 1.8 answer with unit = 3 marks) rk independent of answer)		[3]
	(b)	2.8				[1]
20	(a)	larg	<u>qe</u> fisł	n / fishermen		[1]
	(b)	me abs (sm	rcury sorbeo nall) fi	into water d by micro-organisms sh eat the micro-organisms		[3]
	(c)	bec	cause	they <u>eat</u> fish		[1]